

B.Sc. Semester - 5 (CBCS) Examination

Oct/Nov. - 2018

ORGANIC CHEMISTRY AND SPECTROSCOPY (CORE)

Time: 2:30 Hours

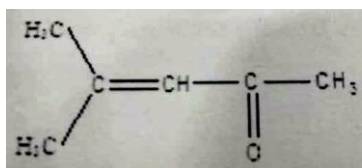
Marks: 70

Instructions:

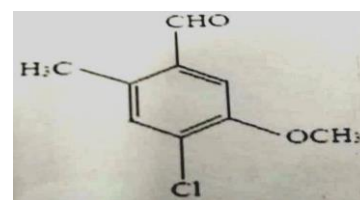
1. All questions are compulsory.
2. Figures to the right indicate marks.

- Que-1 (A) Answer the following question. (04)
Explain step-up reaction (Sowden method).
- (B) Answer any two questions. (10)
(1) Explain determination of ring size of glucose by methylation method.
(2) Explain step-down reaction (Ruff's method).
(3) What is carbohydrate? Discuss the classification of carbohydrates.
- Que-2 (A) Answer the following question. (04)
Give point group of (1) SO_2 (2) $[\text{PtCl}_4]^{2-}$ (3) NH_3 (4) H_3BO_3
- (B) Answer any two questions. (10)
(1) What is plane of symmetry? Explain different types of plane of symmetry with examples.
(2) Prove that in eclipsed ethane $S_n^k \neq E$ but $S_n^{2k} = E$, When n is odd number.
(3) Construct multiplication table for C_{2v} point group.
- Que-3 (A) Answer the following question. (04)
Give preparation and uses of (a) Dulcin (p - phenethyl urea) (b) p - Anisyl urea
- (B) Answer any two questions. (10)
(1) Give preparation and uses of (a) Atenolol (b) Orange-II.
(2) Discuss structure and aromaticity of pyridine.
(3) Give any two synthesis (a) Pyrole (b) Furan (c) Thiophene.
- Que-4 (A) Answer the following question. (04)
Differentiate following pair of compounds using IR spectroscopy,
(a) Propanoic acid and methylethanoate.
(b) P - Methylcyanobenzene and p - aminobenzaldehyde.
- (B) Answer any two questions. (10)
(1) Explain in detail: Fundamental modes of vibrations in IR spectroscopy.
(2) Discuss factors affecting on frequency of $>\text{C}=\text{O}$ band in IR spectroscopy (any five).
(3) Discuss curtiuss rearrangement.
- Que-5 (A) Answer the following question. (04)
Discuss in detail: chromophore and auxochrome.
- (B) Answer any two questions. (10)
(1) Explain constitution of coniine.
(2) Explain Hofmann exhaustive methylation method.
(3) Calculate λ_{max} for the following molecules,

(i)



(ii)



Spectral Data

U.V.:

Empirical rules for Dienes :

	(A) Homoannular $\lambda = 253 \text{ nm.}$ 30nm.	(B) Heteroannular $\lambda = 215 \text{ nm.}$ 30nm.
Increments for double bond extending conjugation		
Exocyclic double bond	5	5
Alkyl substitution or ring residue	5	5
Homocyclic Diene components polar groups:	39	39
- OCOCH ₃	0	0
- OR	6	6
- Cl, -Br	5	5
- NR ₂	60	60

(C) Simple Diene :

Parent $\lambda = 253 \text{ nm.}$

Polar groups:

Alkyl subst for ring

Residue	5 nm
- Cl, -Br	17
- OH	5
- OR	5
- NR ₂	60
- SR	30

(D) Empirical Rules for Enones and Dienones :

(a) Z = C

(1) 6 membered ring or acyclic

(2) 5 membered ring

(b) Z = H

(c) Z = OH or OR

(d) Acyclic dienone

Increment for :

Double bond extending conjugation

Alkyl group of ring residue

				λ
				215
				202
				207
				193
				245
				30
				α 10
				β 12
				18
				5
				39

Polar groups	α	β	λ	δ' other
- Cl	15	12	-	-
- OH	35	30	50	50
- OR	35	30	17	31
- NR ₂	-	93	-	-
- O	-	75	-	-
- NHCOR	-	95	-	-
- OCOCH ₂	6	6	-	6
- SR	-	85	-	-
- Br	25	30	-	-
- NO ₂	-	95	-	-

(E) Empirical Rules for Benzoyl Derivative :

Parent Chromophore :

Z = alkyl or ring residue

Z = H

Mm

246

250

Increment for each substituent

Alkyl or ring residue

-OH; -OCH₃ -OR

- OH

- Cl

- Br

- NH₂- NHCOCH₂- NHCH₃- N(CH₂)₃

0

3

7

11

0

2

13

20

-

20

m

3

7

20

0

2

13

20

-

20

p

10

25

78

10

15

58

45

73

85

Infra – Red Data

Alkene (stretching)

Alkene

Alkyene

Aromatic

Aromatic ring

Alkene

Alkyene

Alkene (Bending)

Aldehyde

Adehyde

Ketone

Carboxylic acid

Ester

Amide

Anhydride

Alcohols, Ethers, esters,

Carboxylic acids, Anydride

Alchols, Phenols :

Free

Bonded

Carboxylic acids

Free

H-bonded

amines (stretch)

Bnding

Nitrile

Ether

Alkene bending

Disulstitued Cis.

Disulstitued Trans.

-C-H

=C-H

=C-H

ArC-H

C=C

>C=C<

-C=C².-C(C₂H₃)₃

-C-H

C=O

C=O

C=O

C=O

C=O

C=O

C-O

2850-2960(v)

3100-3200(m)

3200-3300(s)

3010-3100(m)

1600-1600(v)

(two to three)

1610-1680(v)

2100-2260(s)

1430-1470(m) &

1380-1385(s)

2820-2000(w)&2650 2760(s)

1740-1720(s)

1725-1710(s)

1725-1705(s)

1750-1730(s)

1670-1640(s)

1810-1860(s)&1740-1790

1800-1000(s)

O-H

O-H

O-H

O-H

N-H

-N-H

-C=N

-O-

3650-3600(sh)

3500-3200(b)

3500-3650(m)

2500-3200(b)

1640-1550(m)

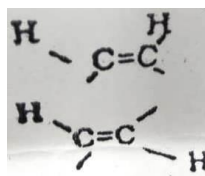
1640-1550(m)

2210-2280(s)

1070-1150(s)

-690(s)

960-970(s)



Aromatic substitution:

Type C-H out of plane bending

No. of adjacent H atom.

5

4

3

2

1

Range cm
750(s) & 700(s)

750

780

830

850

